Lights

Astronomy is based on

observing lights from celestial bodies.

And God said, "Let there be light"; and there was light. And God saw that the light was good; (Bible: Genesis I)



Cat's Eye planetary Nebula

Photons: "bullets" that hit our retina ---- "bullets" with color

Leaf bobs up and down with

nebble thrown into pond

frequency of wave.

peak

trough

wavelength

speed of wave moving outward

A photon is both a particle and a wave *(wave-particle duality)* so it has a **wavelength** associated with it.

So is an electron, a proton.... a human, a galaxy...

shorter wavelength || higher energy

a radio photon (~ metre) ~ 10^{-6} electron volts

visible photon (~ $5x10^{-7}$ m) ~ 2 electron volts

X-ray photon (~ 10^{-9} m) ~ 1000 electron volts *able to wreak havoc..* Color of a Photon?

--- Color is in the eyes of the beholder.

photons are only tagged with wavelengths human eye converts this information to 'color'



longer wavelength photon look redder

shorter: look bluer

Visible light





What is color WHITE?

Defined to the composite color of the Sun

The Sun's chromaticity (seen from space, CIE x y: 0.326 0.338





Here is a comparison with various whi	itepoints:		
1	x y	sRGB pixel color	
Sun above atmosphere	0.3259 0.3379	#fff3ea	
Illuminant B ("direct sunlight")	0.3840 0.3516	#ffbfaa	+blue sky=white lig



The electro-magnetic radiation spectrum

THE ELECTROMAGNETIC SPECTRUM





What makes the **color** of an object?



Most objects around us are seen by their reflected light --- Relies on external illumination.

All objects also emit light.

Candle ~ 3000K, visible

 $Cat \sim 300K$, infrared







Cosmos: 3 Kelvin background, radio

So each object around us has really two colors.

- 1) it reflects light shine on it. Its material property (wood, steel, paper...) determines this color.
- 2) it also radiates photons because it is not 0 Kelvin. Its temperature determines the color of this radiation.

Astronomical objects are mostly seen by their radiation. Namely, you see them because they are hot. Their color therefore depends on their temperature (only).

We call them 'blackbody' because they are as if they don't reflect lights.

Universal Radiation of a **blackbody** (only depends on T):



Color of stars



Blue stars: ~ 20,000K red stars: ~ 2,000 K

NGC 2266 star cluster

. . .

"true color" of an object

(seen in emitted light)



Spectrum



Spectrum of the Sun

Spectrum: dispersed light according to wavelength



NSO Solar telescope





Mid-resolution solar spectrum

dark lines: deficit of flux

and a really high resolution solar spectrum...





Wavelengths of the spectral features correspond to energy differences between different states.The states depend on what atom it is.

Low --> High, absorb a photon High --> Low, releases a photon

Hydrogen spectra



Spectrum of a star:

- 1) smooth blackbody radiation from the hot interior,
- 2) cooler gas near the surface absorb at particular frequencies
 - --- forming absorption lines (absorption spectrum) *"silouette of a person in front of bright car light"*
- 3) hotter gas near the surface (corona) emits in certain frequencies
 - --- emission spectrum





sodium





Optical Spectrum of the Solar lights ---- Fraunhofer lines

from www.harmsy.freeuk.com/fraunhofer.html

absorption features: flux deficit in the radiation spectrum



Chemical composition of the celestial objects is revealed.

I. One example of motion in astronomy projected positions on the celestial orb changes



II. but stars/galaxies also move toward or away from us. The projected position doesn't change.

So how do we know that they are moving?



Finding out velocities for any moving objects

- 1) Expansion of the universe
- 2) detecting planets around other stars
- 3) finding blackholes
- 4) measuring stellar masses in binaries
- 5) rotation of the galaxy
- ... getting out of the way for the fire-engine





you

Color representation in astronomical images bluer:shorter wavelength redder: longer wavelength



"false-color image"





Sombrero Galaxy