

Physical Ingredients for Constructing a Star (or Planet)

1. Support against gravity

Pressure: hydrostatic equilibrium
equation of state

Energy: virial theorem

2. Source of energy

Contraction

Nuclear power

Photon propagation/emission

Nuclear Fusion

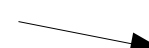
Einstein: $E = m c^2$

atomic unit: $u = 1.660540 \times 10^{-27} \text{ kg}$ (mass of $^{12}\text{C}/12$)

proton: $m_p = 1.672623 \times 10^{-27} \text{ kg} = 1.0072765 u = \mathbf{938.79 \text{ MeV}}$

neutron: $m_n = 1.674929 \times 10^{-27} \text{ kg} = 1.0086653 u$

electron: $m_e = 9.109390 \times 10^{-31} \text{ kg} = 0.0054858 u$

Hydrogen atom: $m_H = 1.007825 u = m_p + m_e - \text{electro-static}/2$  13.6 eV binding energy of H atom

Helium nucleus: $m_{\text{He}} = 4.00151 u = 2 m_p + 2 m_n - \Delta m$
 $\Delta m = 0.03037 u \sim 0.7\% (4 m_H) \sim 28 \text{ MeV}$

Fusion $4 p \rightarrow \text{He}$ releases $\sim 28 \text{ MeV}$

Binding Energy of a Nucleus

--- energy released when formed
 --- energy required to unbind

${}^1_1\text{H}$ $E_b = 0$

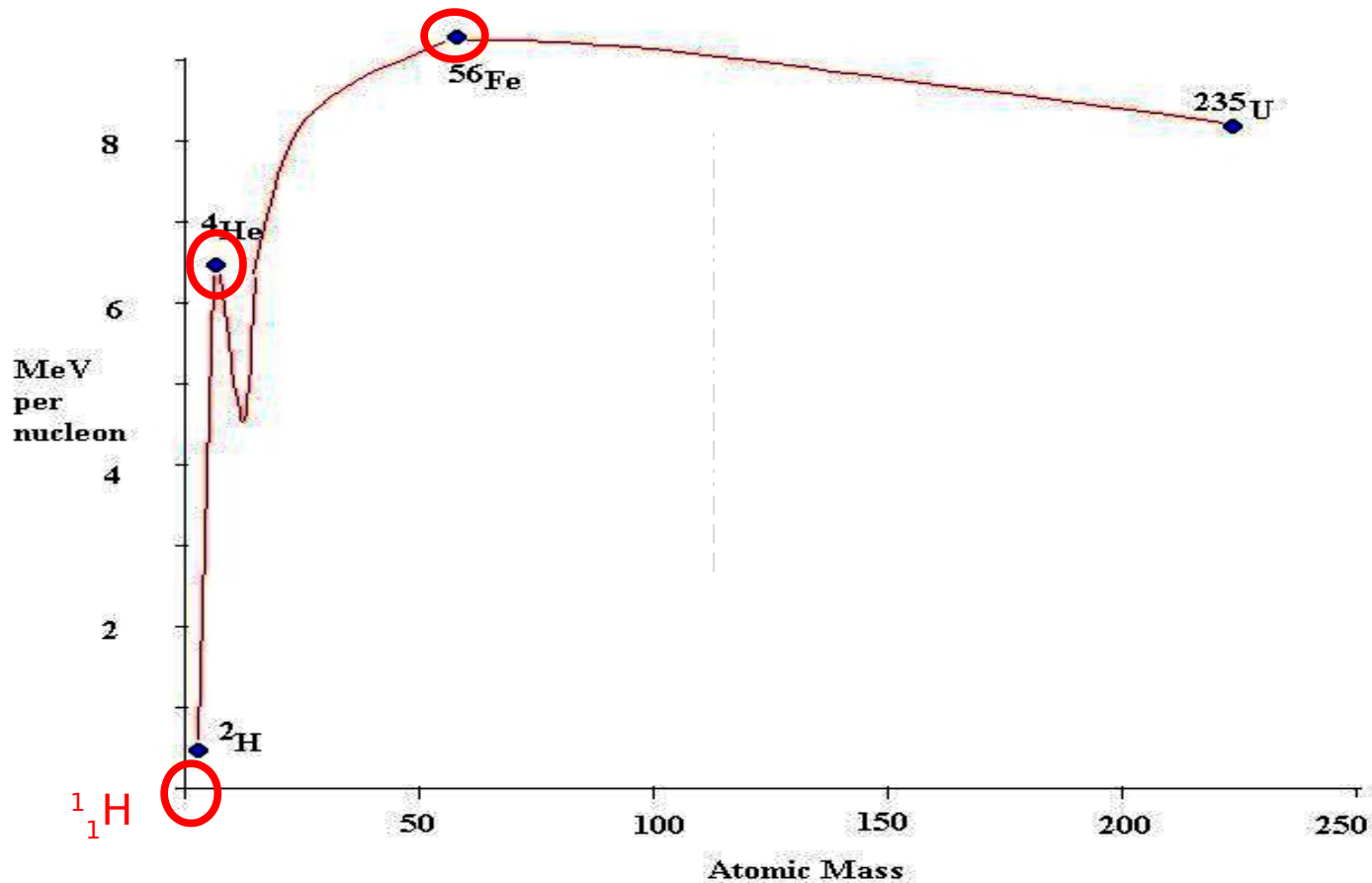
${}^4_2\text{He}$ $\Delta m c^2 = 28.03 \text{ MeV}$, hence 7.08 MeV/nucleon

${}^{16}_8\text{O}$ $(Z m_p + (A-Z) m_n - m_{\text{nucleus}}) c^2 / A =$ 7.97 MeV/nucleon

${}^{56}_{26}\text{Fe}$ 8.79 MeV/nucleon

$({}^{238}_{92}\text{U})$ 7.3 MeV/nucleon

Most stable nucleus
 easily made in stars.



Basic concepts in nuclear physics

1) Binding energy of an element: A_ZX : $E_b = (Z m_p + (A-Z) m_n - m_{\text{nucleus}}) c^2$

2) attractive **strong force** binds protons & neutrons together against repulsive Coulomb force

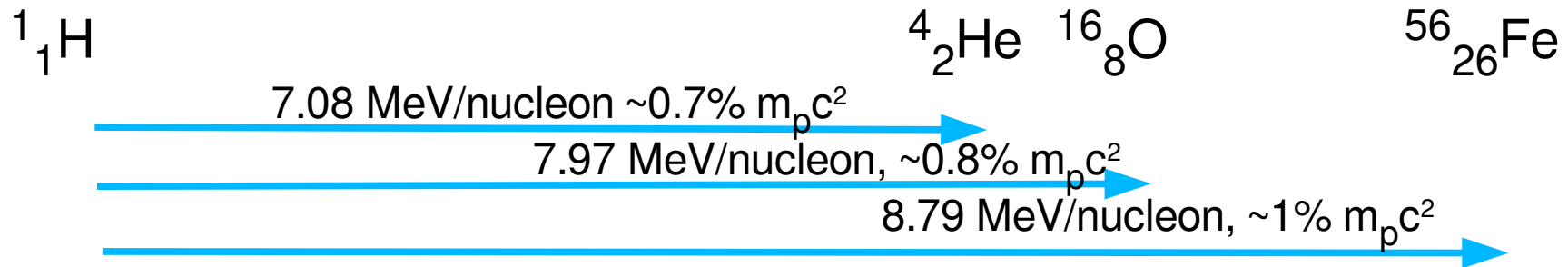
Four basic forces in nature

long-range forces: gravity, electro-magnetism (force carriers have no mass, $F \sim 1/r^2$)
short-range forces: strong force, weak force (force carriers have mass,
strength falls off dramatically beyond nuclear dimension $\sim 1 \text{ fm} = 10^{-15} \text{ m}$)

3) For A less than a critical value (~ 56),
strong force increases with A faster than Coulomb repulsion
 \Rightarrow binding/nucleon increases with A
Beyond that, nucleus too large \Rightarrow binding/nucleon decreases A

---- energetically favourable to fuse H to He to Fe, but not past Fe
---- past Fe, fission energetically favourable; fusion becomes endothermic

Nuclear Energy Yield: H burning to Fe yields ~ 9 MeV/nucleon



Fusion: Each proton can maximally yield $\sim 1\% m_p c^2 \sim 9 \text{ MeV} \sim 10^{-12} \text{ J}$
(1 g of Hydrogen fusion ~ annual energy consumption of one Canadian)

Fusion energy available for Sun: $E \sim 10^{-12} \text{ J} \times M_\odot / m_p \sim 10^{45} \text{ J} \gg GM_\odot^2 / R_\odot$

Nuclear Timescale: $t_{\text{nuc}} \sim E / L_\odot \sim 10^{45} \text{ J} / L_\odot \sim 10^{11} \text{ yr}$

Actual lifespan $\sim 10^{10}$ years (now about half-way)

- 1) L increases in later life;
- 2) not all H burned;
- 3) not burned to Fe.

Compare: dynamical time: $t_{\text{dyn}} \sim 30 \text{ min}$
 thermal time: $t_{\text{KH}} \sim 10^7 \text{ yr}$

Reminder: Periodic Table of the Elements

(from <http://www.chemcool.com>)

H ¹																	He ²
Li ³	Be ⁴											B ⁵	C ⁶	N ⁷	O ⁸	F ⁹	Ne ¹⁰
Na ¹¹	Mg ¹²											Al ¹³	Si ¹⁴	P ¹⁵	S ¹⁶	Cl ¹⁷	Ar ¹⁸
K ¹⁹	Ca ²⁰	Sc ²¹	Ti ²²	V ²³	Cr ²⁴	Mn ²⁵	Fe ²⁶	Co ²⁷	Ni ²⁸	Cu ²⁹	Zn ³⁰	Ga ³¹	Ge ³²	As ³³	Se ³⁴	Br ³⁵	Kr ³⁶
Rb ³⁷	Sr ³⁸	Y ³⁹	Zr ⁴⁰	Nb ⁴¹	Mo ⁴²	Tc ⁴³	Ru ⁴⁴	Rh ⁴⁵	Pd ⁴⁶	Ag ⁴⁷	Cd ⁴⁸	In ⁴⁹	Sn ⁵⁰	Sb ⁵¹	Te ⁵²	I ⁵³	Xe ⁵⁴
Cs ⁵⁵	Ba ⁵⁶	La ⁵⁷	Hf ⁷²	Ta ⁷³	W ⁷⁴	Re ⁷⁵	Os ⁷⁶	Ir ⁷⁷	Pt ⁷⁸	Au ⁷⁹	Hg ⁸⁰	Tl ⁸¹	Pb ⁸²	Bi ⁸³	Po ⁸⁴	At ⁸⁵	Rn ⁸⁶
Fr ⁸⁷	Ra ⁸⁸	Ac ⁸⁹	Rf ¹⁰⁴	Db ¹⁰⁵	Sg ¹⁰⁶	Bh ¹⁰⁷	Hs ¹⁰⁸	Mt ¹⁰⁹	Uun ¹¹⁰								

Ce ⁵⁸	Pr ⁵⁹	Nd ⁶⁰	Pm ⁶¹	Sm ⁶²	Eu ⁶³	Gd ⁶⁴	Tb ⁶⁵	Dy ⁶⁶	Ho ⁶⁷	Er ⁶⁸	Tm ⁶⁹	Yb ⁷⁰	Lu ⁷¹
Th ⁹⁰	Pa ⁹¹	U ⁹²	Np ⁹³	Pu ⁹⁴	Am ⁹⁵	Cm ⁹⁶	Bk ⁹⁷	Cf ⁹⁸	Es ⁹⁹	Fm ¹⁰⁰	Md ¹⁰¹	No ¹⁰²	Lr ¹⁰³

Uranium & Plutonium

Elements used by the Human body

H ¹																	He ²
Li ³	Be ⁴											B ⁵	C ⁶	N ⁷	O ⁸	F ⁹	Ne ¹⁰
Na ¹¹	Mg ¹²											Al ¹³	Si ¹⁴	P ¹⁵	S ¹⁶	Cl ¹⁷	Ar ¹⁸
K ¹⁹	Ca ²⁰	Sc ²¹	Ti ²²	V ²³	Cr ²⁴	Mn ²⁵	Fe ²⁶	Co ²⁷	Ni ²⁸	Cu ²⁹	Zn ³⁰	Ga ³¹	Ge ³²	As ³³	Se ³⁴	Br ³⁵	Kr ³⁶
Rb ³⁷	Sr ³⁸	Y ³⁹	Zr ⁴⁰	Nb ⁴¹	Mo ⁴²	Tc ⁴³	Ru ⁴⁴	Rh ⁴⁵	Pd ⁴⁶	Ag ⁴⁷	Cd ⁴⁸	In ⁴⁹	Sn ⁵⁰	Sb ⁵¹	Te ⁵²	I ⁵³	Xe ⁵⁴
Cs ⁵⁵	Ba ⁵⁶	La ⁵⁷	Hf ⁷²	Ta ⁷³	W ⁷⁴	Re ⁷⁵	Os ⁷⁶	Ir ⁷⁷	Pt ⁷⁸	Au ⁷⁹	Hg ⁸⁰	Tl ⁸¹	Pb ⁸²	Bi ⁸³	Po ⁸⁴	At ⁸⁵	Rn ⁸⁶
Fr ⁸⁷	Ra ⁸⁸	Ac ⁸⁹	Rf ¹⁰⁴	Db ¹⁰⁵	Sg ¹⁰⁶	Bh ¹⁰⁷	Hs ¹⁰⁸	Mt ¹⁰⁹	Uun ¹¹⁰								

Ce ⁵⁸	Pr ⁵⁹	Nd ⁶⁰	Pm ⁶¹	Sm ⁶²	Eu ⁶³	Gd ⁶⁴	Tb ⁶⁵	Dy ⁶⁶	Ho ⁶⁷	Er ⁶⁸	Tm ⁶⁹	Yb ⁷⁰	Lu ⁷¹
Th ⁹⁰	Pa ⁹¹	U ⁹²	Np ⁹³	Pu ⁹⁴	Am ⁹⁵	Cm ⁹⁶	Bk ⁹⁷	Cf ⁹⁸	Es ⁹⁹	Fm ¹⁰⁰	Md ¹⁰¹	No ¹⁰²	Lr ¹⁰³

How to set the Nuclear Fire?

(or why we don't yet have clean fusion power on Earth)



Problem 1: Coulomb barrier

protons have to **overcome the electrostatic repulsion** between them and reach the realm of strong force
($\sim 1 \text{ fm} = 10^{-15} \text{ m}$, also the size of a nucleus)

(e^- too far out to be relevant, $\sim 10^{-11} \text{ m}$)

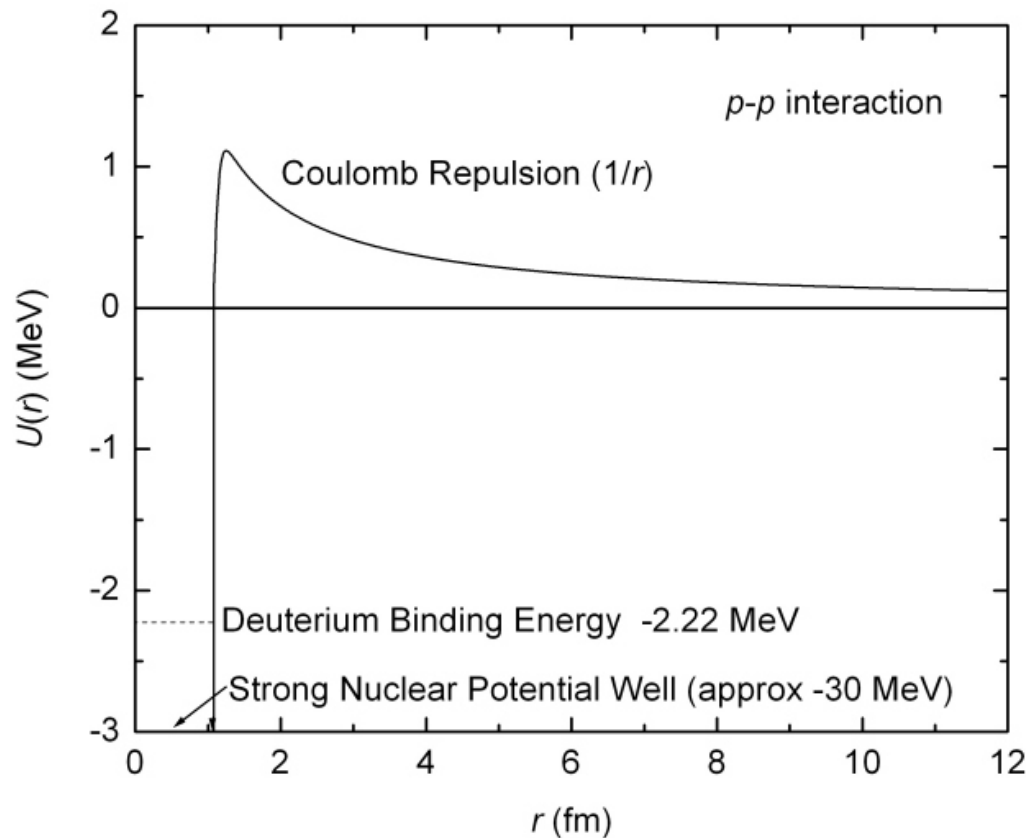
Classical solution: thermal motion of the nuclei

Two protons need to get as close as ~ 1 fm for strong force attraction to set in

Coulomb Force is a potential force (like gravity, but can repel)

$$\text{Coulomb potential } U_c = \frac{1}{4\pi\epsilon_0} \frac{e_1 e_2}{r} \quad (1.4 \text{ MeV for } r=1 \text{ fm})$$

As two protons approach each other, total energy is conserved:



$$\frac{1}{2} m_p v_\infty^2 + U_c(\infty) = U_c(1 \text{ fm})$$

$$\frac{1}{2} m_p v_\infty^2 = \frac{3}{2} k T \geq \frac{e^2}{4\pi\epsilon_0(1 \text{ fm})}$$

$$T \geq T_{\text{classical}} = \frac{e^2}{6k\pi\epsilon_0(1 \text{ fm})} \sim 10^{10} \text{ K}$$

But from Virial Theorem,
we know that Sun has $T_c \sim 10^7 \text{ K}$

Sir Arthur Eddington (1882-1944)



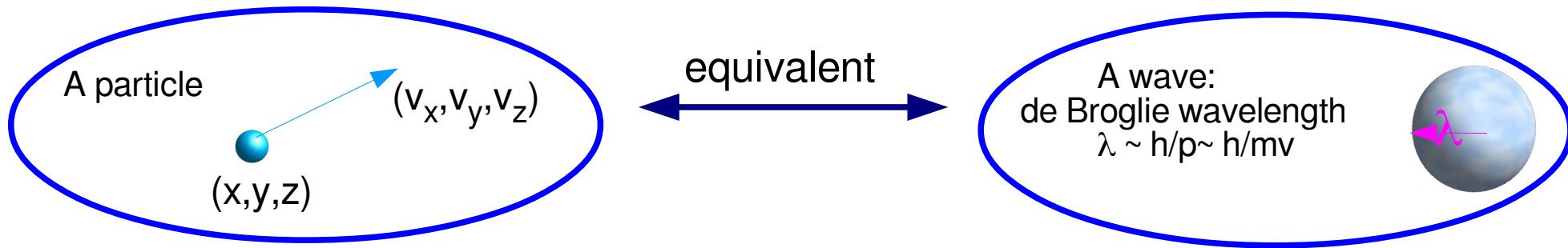
Arthur Eddington thought that nuclear processes must be involved to account for the radiant energy of the sun, but was criticized because the temperature was seen to be not hot enough when considered by classical physics alone.

His tongue-in-cheek reply to his critics: *"I am aware that many critics consider the stars are not hot enough. The critics lay themselves open to an obvious retort; we tell them to go and find a hotter place."*

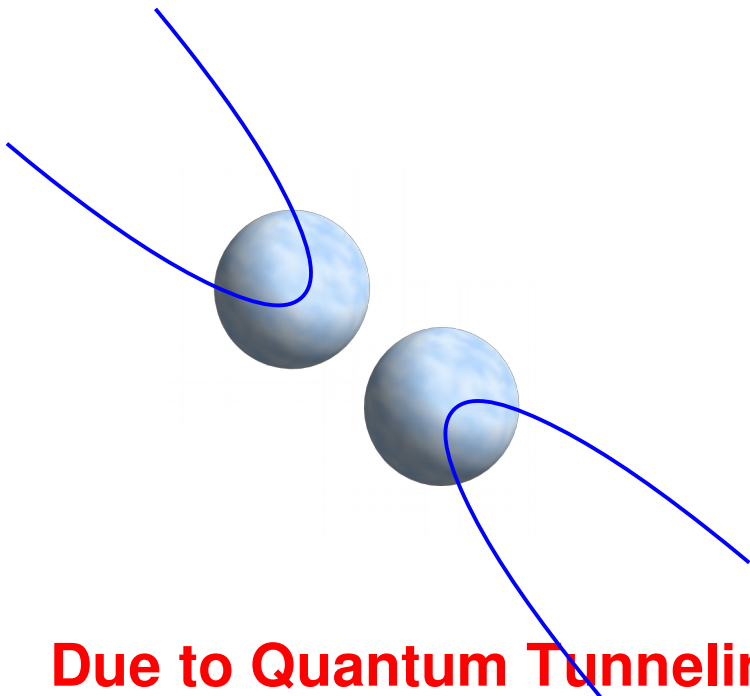
**Vindicated by quantum mechanics.
Protons are 'BIG'.**

Quantum mechanical solution: Wave-Particle Duality

(size of a particle depends on its momentum)



Use *Quantum Tunneling*: if a proton gets within one de Broglie wavelength of another, there is a certain probability that they “are” closer than ~ 1 fm



$$U_c = \frac{e^2}{4\pi\epsilon_0\lambda} \leq \frac{1}{2}m_p v_\infty^2 = \frac{1}{2} \frac{p^2}{m_p} \sim \frac{1}{2m_p} \left(\frac{h}{\lambda}\right)^2$$

$$\lambda \leq \frac{4\pi\epsilon_0 h^2}{2e^2 m_p} \sim 10^{-13} m \sim 100 \text{ fm}$$

$$\frac{3}{2}kT = \frac{1}{2}m_p v_\infty^2 = \frac{1}{2m_p} \left(\frac{h}{\lambda}\right)^2$$


$$T \geq T_{\text{quantum}} \sim \frac{m_p e^4}{12\pi^2 \epsilon_0^2 k h^2} \sim 10^7 \text{ K}$$

Due to Quantum Tunneling, hydrogen fusion is possible already at 10^7 K
 (stars $< 0.08M_\odot$ cannot reach this temperature at the center; “failed stars” or brown dwarfs)

Quantum tunneling and the ignition temperature


Classical Picture


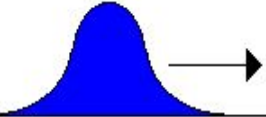
electron  → electric field 

 ← in classical physics, the electron is repelled by an electric field as long as energy of electron is below energy level of the field

Quantum Picture

electron wave  → 

 in quantum physics, the wave function of the electron encounters the electric field, but has some finite probability of tunneling through

  →

Quantum Tunneling:
a physical system spends
some time in an energetically
forbidden region:

*there always exists
a certain probability
that two protons 'are'
closer than $\sim 1\text{fm}$*

Tunneling probability

$$\exp(-2\pi^2 U_c(\lambda(E))/E)$$

$$\sim 4 \times 10^{-11} \quad (E/k \sim T \sim 10^7 \text{ K})$$

$$\sim 0.4 \quad (E/k \sim T \sim 10^{10} \text{ K})$$

Nuclear reaction rates rise
extremely steeply with T.
Hence, **ignition temperature**
($\sim 10^7 \text{ K}$ for Hydrogen fusion)

How to set the Nuclear Fire? (cont'd)

Problem 2: How often do protons see each other?

- 1) space is empty for nuclei even at the Sun's center ($\rho \sim 100 \text{ g/cm}^3$)
mean separation $\sim 10^{-11} \text{ m} \gg \lambda \sim 10^{-13} \text{ m}$
- 2) $4 \text{ }^1_1\text{H} \rightarrow \text{}^4_2\text{He} + \text{energy release of } \sim 7 \text{ MeV/nucleon}$
All four in the same place at the same time? --- difficult
Fusion proceeds through *a chain of 2-body reactions*,
each satisfying conservation of energy, momentum, charge,
lepton number (e^- , e^+ , ν , anti- ν), and baryon number (A)
 - a) At low T: **p-p chain** (involving only protons and products);
 - b) At higher T: **CNO chain** (C,N,O as catalysts).
- 3) The more highly charged the nucleus, the higher the Coulomb barrier,
and thus the higher the required temperature;
Helium fusion requires $T > 10^8 \text{ K}$; others $T > 10^9 \text{ K}$
(*relevant after H exhaustion in the stellar core*)